

# IT'S ABOUT TIME

Characteristic times in biogeochemistry and climate

Stephen E. Schwartz



Symposium on biogeochemical cycling and climate



In honor of Henning Rodhe on the occasion of his retirement from  
the chair of Chemical Meteorology

Stockholm University

May 23, 2008

# IT'S ABOUT TIME



## TOPICS

Turnover times of atmospheric aerosols

Turnover time of excess atmospheric CO<sub>2</sub>

Characteristic time of Earth's climate system

# TURNOVER TIMES OF ATMOSPHERIC AEROSOLS

I shot an *aerosol particle* into the air, and then  
It fell to earth, I knew not where (*or when*);

*Apologies to Longfellow*

Source-receptor relations in acid deposition

Aerosol influences on climate

# A note on the concepts of age distribution and transit time in natural reservoirs

By BERT BOLIN and HENNING RODHE, *Institute of Meteorology,  
University of Stockholm, Sweden*<sup>1</sup>

Tellus XXV (1973)

reservoirs. We shall further limit the present study to steady state conditions, i.e. we assume that the total mass and the statistical distributions studied do not vary with time. Such a

# Participants at the International Symposium on Sulfur in the Atmosphere, Dubrovnik, September 1977





# BUDGETS AND TURN-OVER TIMES OF ATMOSPHERIC SULFUR COMPOUNDS

HENNING RODHE

*Atmospheric Environment* Vol. 12, 1978.

*Steady-state conditions:*

Turnover time:  $t_{-o} = M / F$

$M$  = mass in reservoir

$F$  = flux into (or out of) reservoir

*Turnover times:*

SO<sub>2</sub>                      25 h

SO<sub>4</sub><sup>2-</sup>                      80 h

Sulfur (SO<sub>2</sub>+ SO<sub>4</sub><sup>2-</sup>) 50 h

*Sulfate yield: 30%*

# **Residence times in reservoirs under non-steady-state conditions: application to atmospheric SO<sub>2</sub> and aerosol sulfate**

By STEPHEN E. SCHWARTZ, *Atmospheric Sciences Division, Brookhaven National Laboratory, Upton, New York 11973, U.S.A.*

*Tellus* (1979)

*Non-steady-state conditions:*

$$\text{Turnover time: } \tau_{t-o} = \bar{M} / \bar{F}$$

$M$  = mass in reservoir

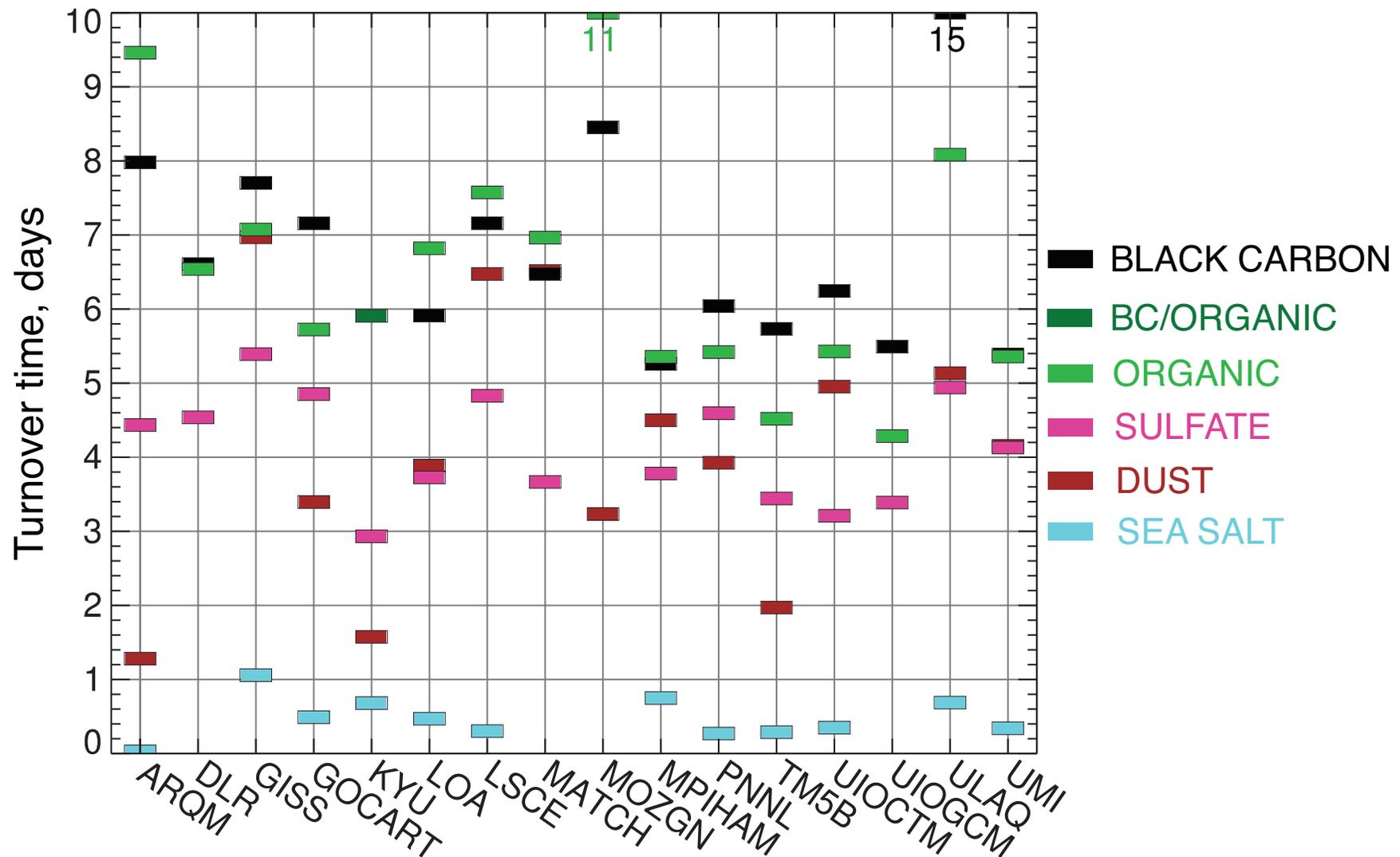
$F$  = flux *out of* reservoir

Averages are taken over suitably long periods.

Long-time averages yield stable, representative turnover times.

# TURNOVER TIME AS INTENSIVE VARIABLE

Five aerosol species in 16 global chemical transport models



AEROCOM, Textor et al., ACP, 2006

Characterize and compare processes in chemical transport models.

Turnover time displays *wide model-to-model variance*.

Sulfate residence time close to Rodhe (1978) value 80 h = 3.3 days

# TURNOVER TIME OF EXCESS ATMOSPHERIC CO<sub>2</sub>

I emit a kilogram of CO<sub>2</sub> into the air, and then

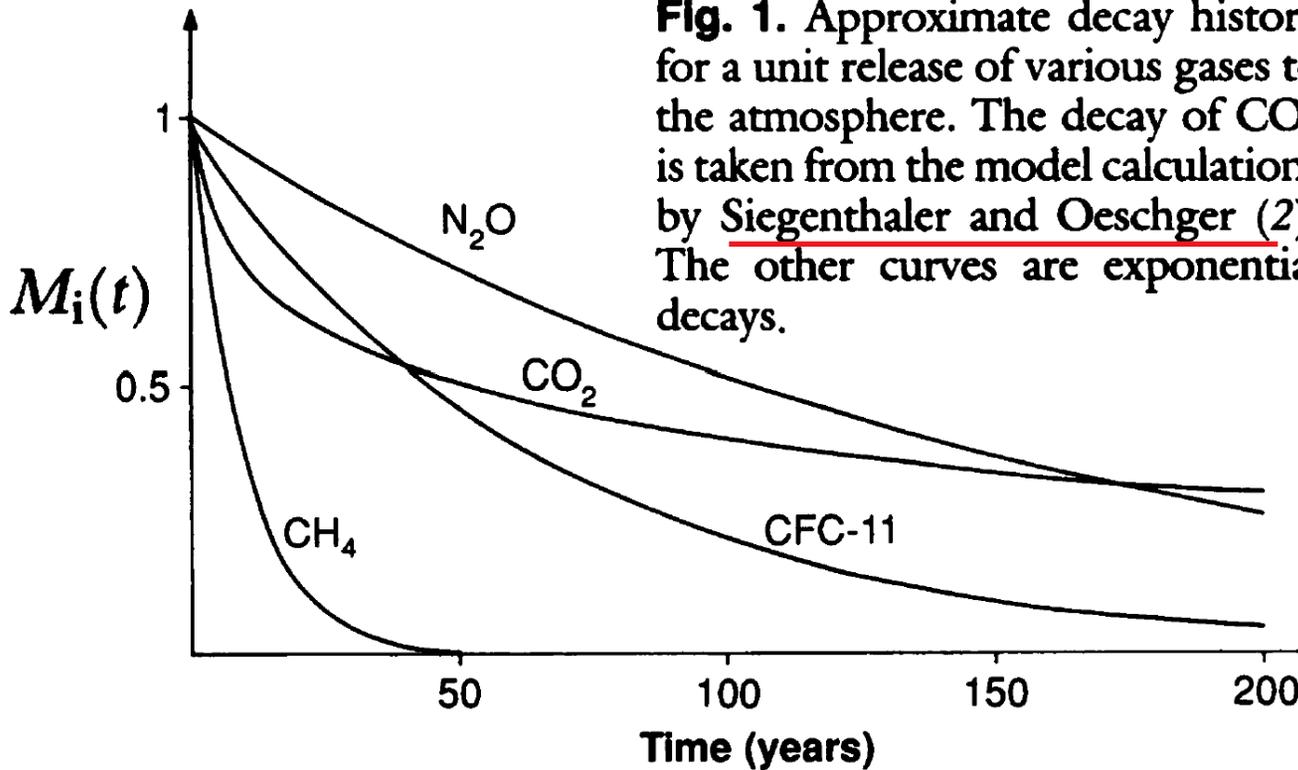
It falls to earth: Can you tell me when?

Decision making on energy and climate change

# A Comparison of the Contribution of Various Gases to the Greenhouse Effect

HENNING RODHE SCIENCE, VOL. 248 8 JUNE 1990

Impulse  
response  
function



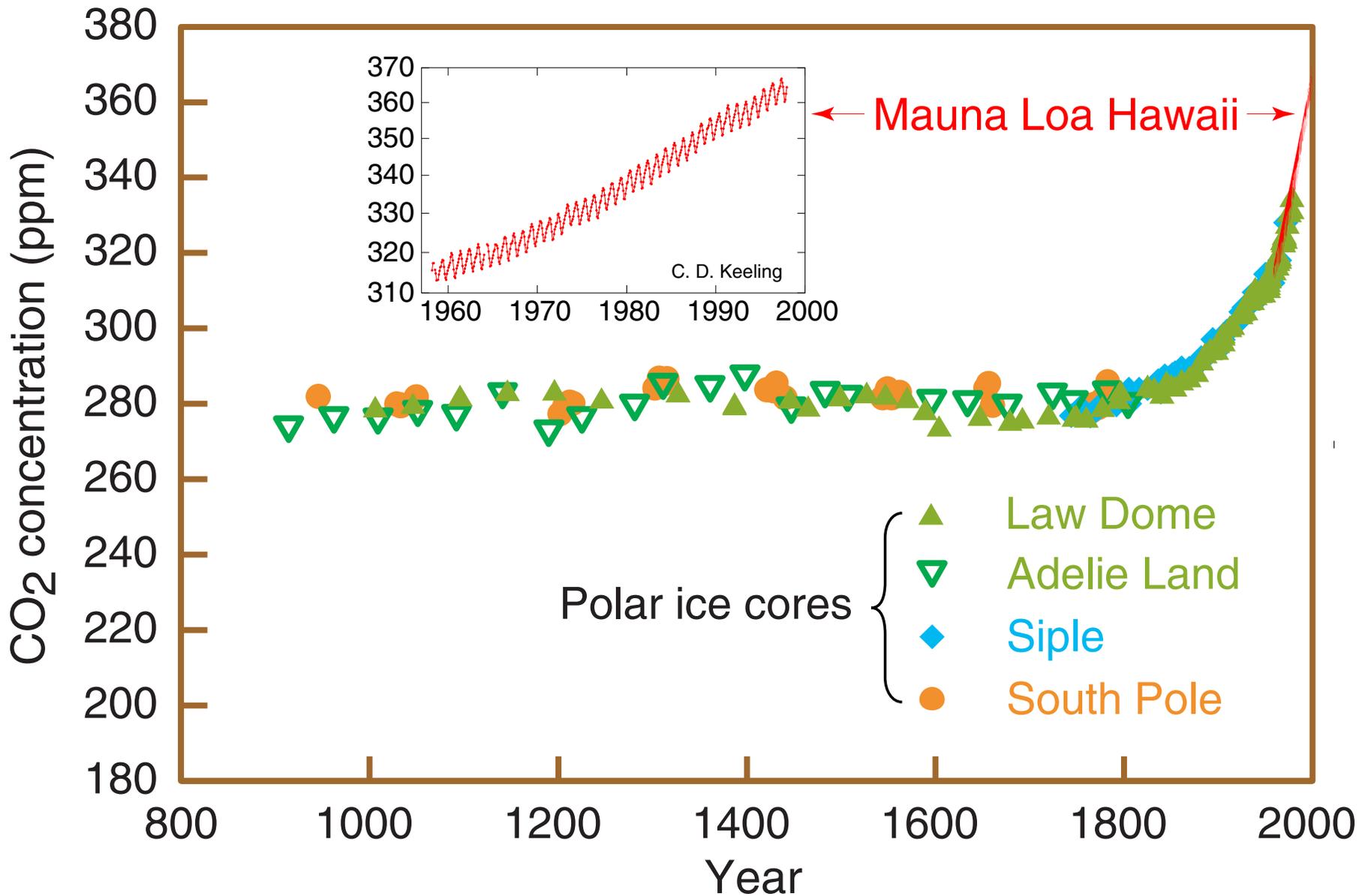
**Fig. 1.** Approximate decay history for a unit release of various gases to the atmosphere. The decay of  $CO_2$  is taken from the model calculations by Siegenthaler and Oeschger (2). The other curves are exponential decays.

$$AG_i = \int_0^T k_i M_i(t) dt = k_i \int_0^T M_i(t) dt$$

Forcing  
per unit  
mass

The question arises what time scale  $T$  one should use in this integration.

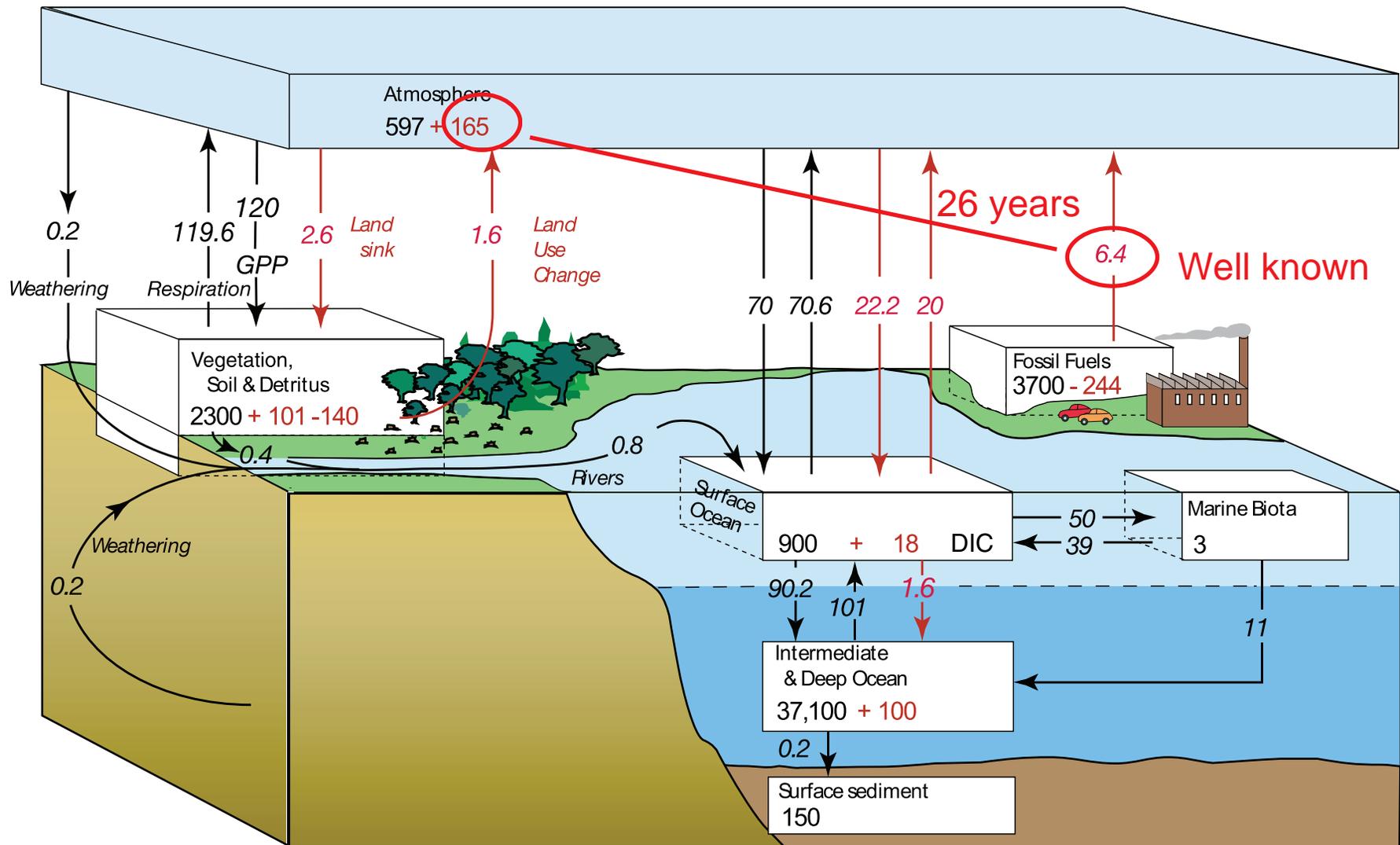
# ATMOSPHERIC CARBON DIOXIDE IS INCREASING



Global carbon dioxide concentration over the last thousand years

# THE GLOBAL CARBON CYCLE

Preindustrial and **anthropogenic perturbation (1990's)**  
 Stocks in upright type, Pg C; *flows in italic type*, Pg C yr<sup>-1</sup>

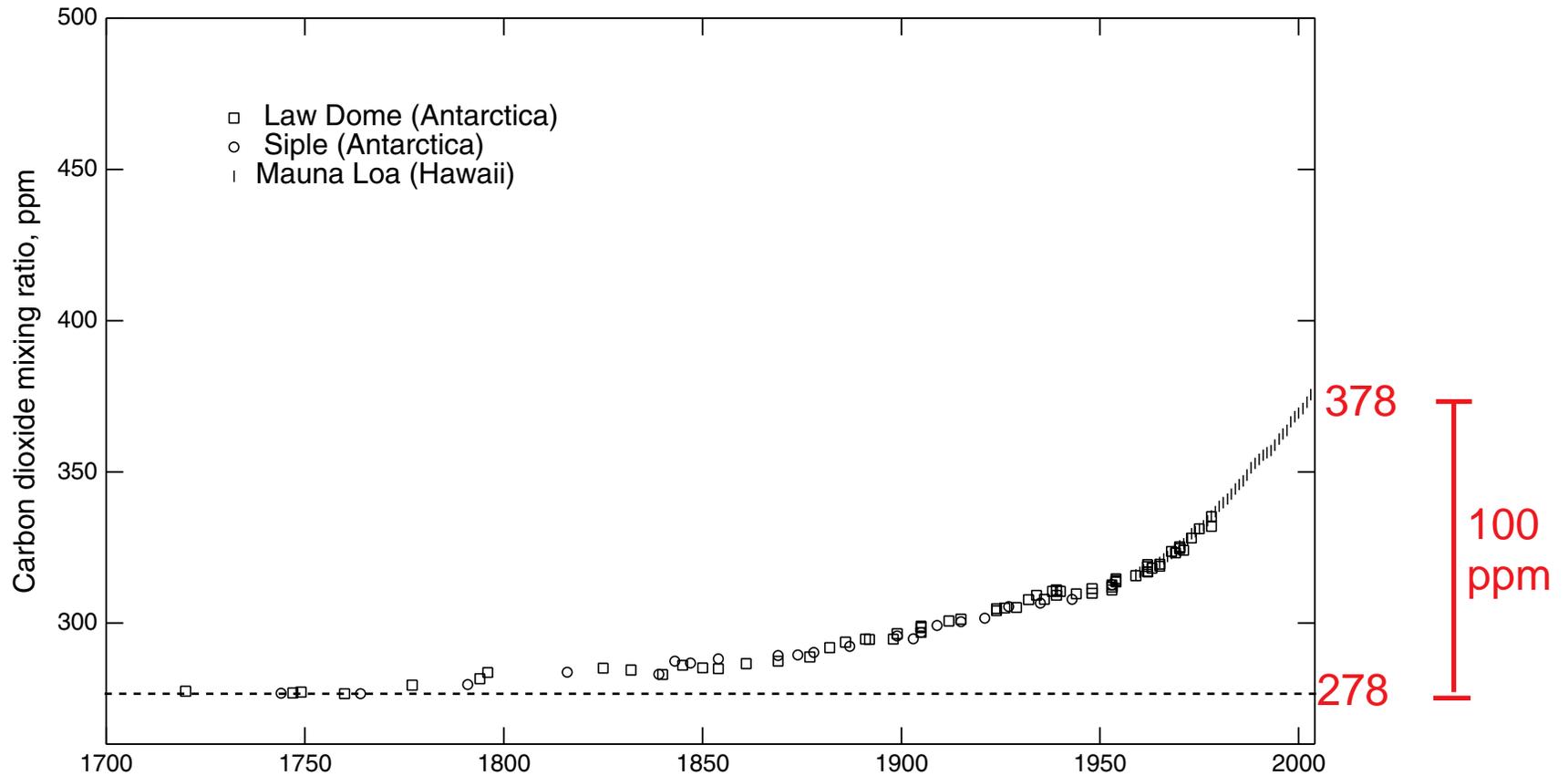


IPCC AR4, Chapter 7 (2007); after Sarmiento and Gruber (2002)

**Excess atmospheric carbon dioxide is 26 years of current fossil fuel emissions.**

# ATMOSPHERIC CARBON DIOXIDE

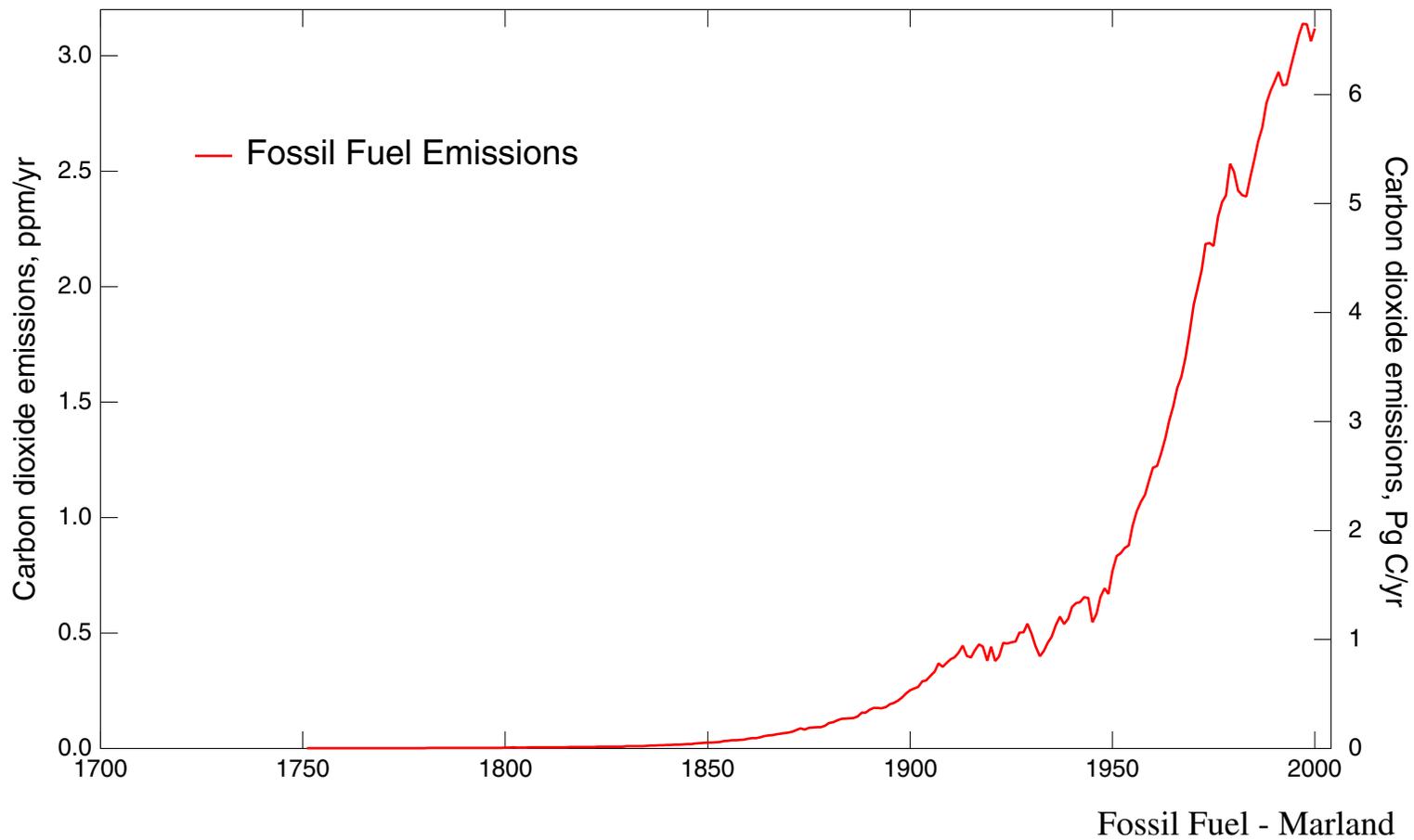
## Time series 1700 - 2003



Law - Etheridge et al.  
Siple - Friedli et al.  
Mauna Loa - Keeling

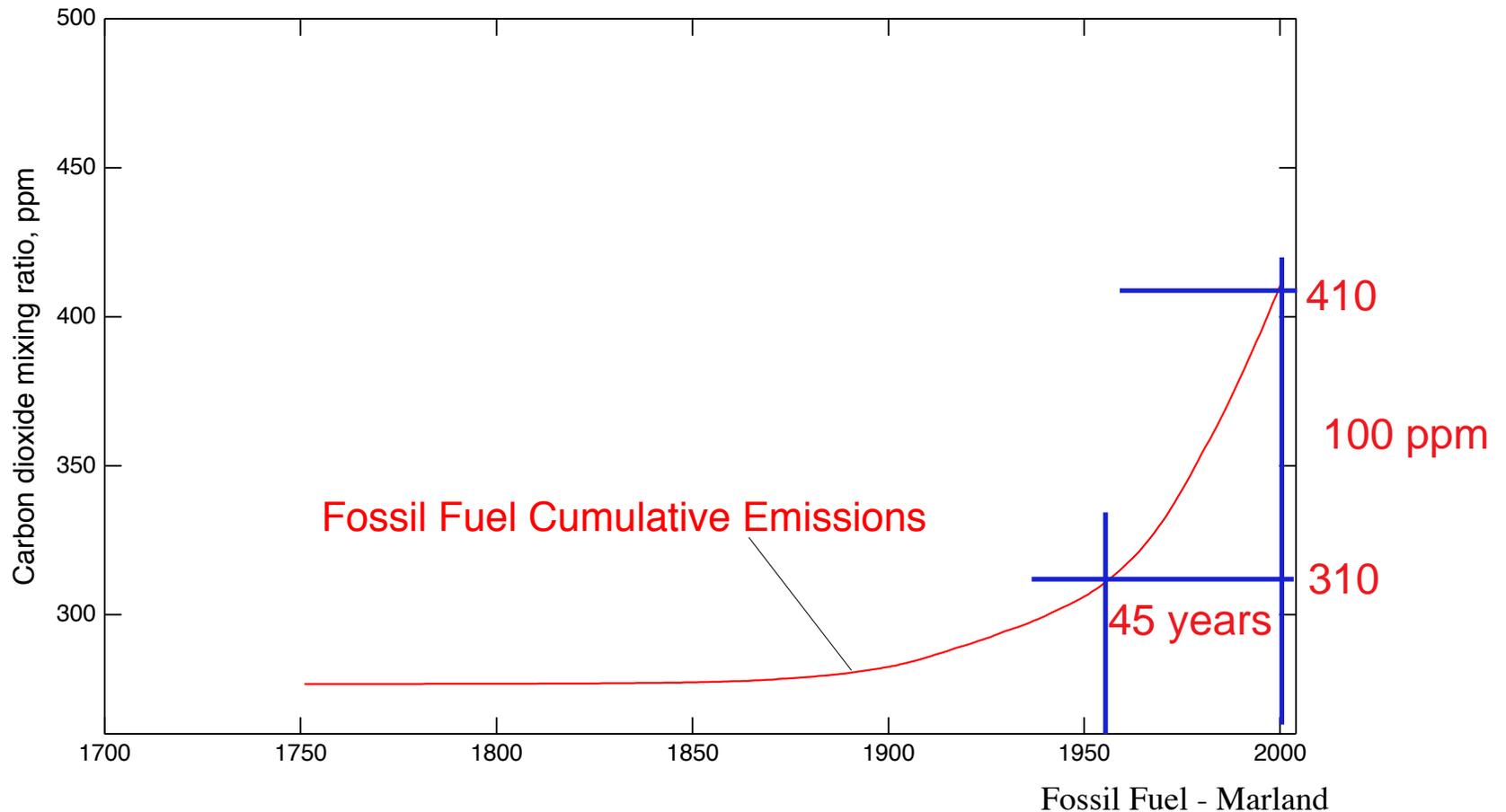
# FOSSIL FUEL CO<sub>2</sub> EMISSIONS

## Time series 1700 - 2003



# FOSSIL CO<sub>2</sub> CUMULATIVE EMISSIONS

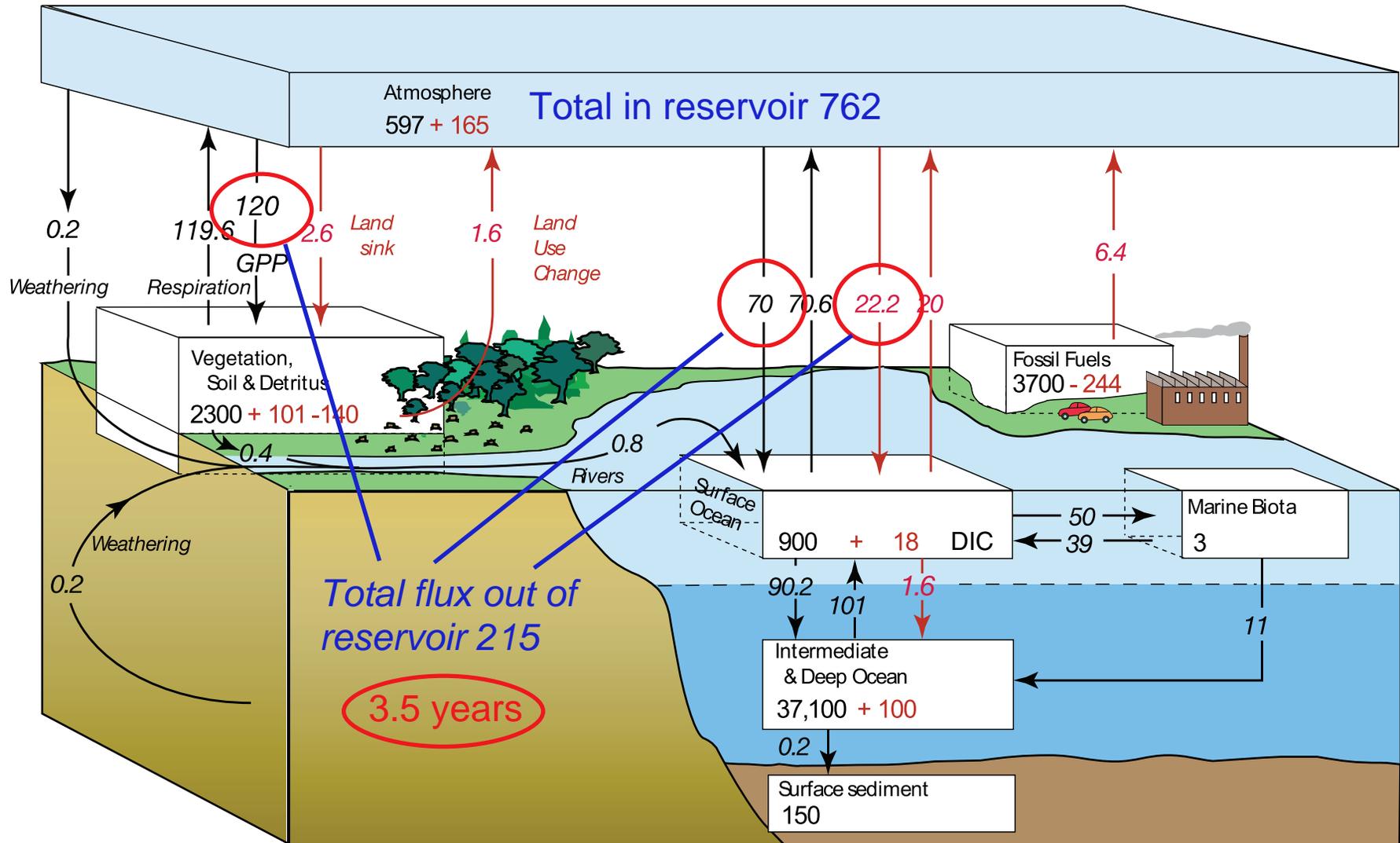
Time series 1700 - 2003



Present excess CO<sub>2</sub> in the atmosphere is equal to 45 years of fossil fuel emissions.

# THE GLOBAL CARBON CYCLE

Preindustrial and **anthropogenic perturbation (1990's)**  
 Stocks in upright type, Pg C; *flows in italic type, Pg C yr<sup>-1</sup>*

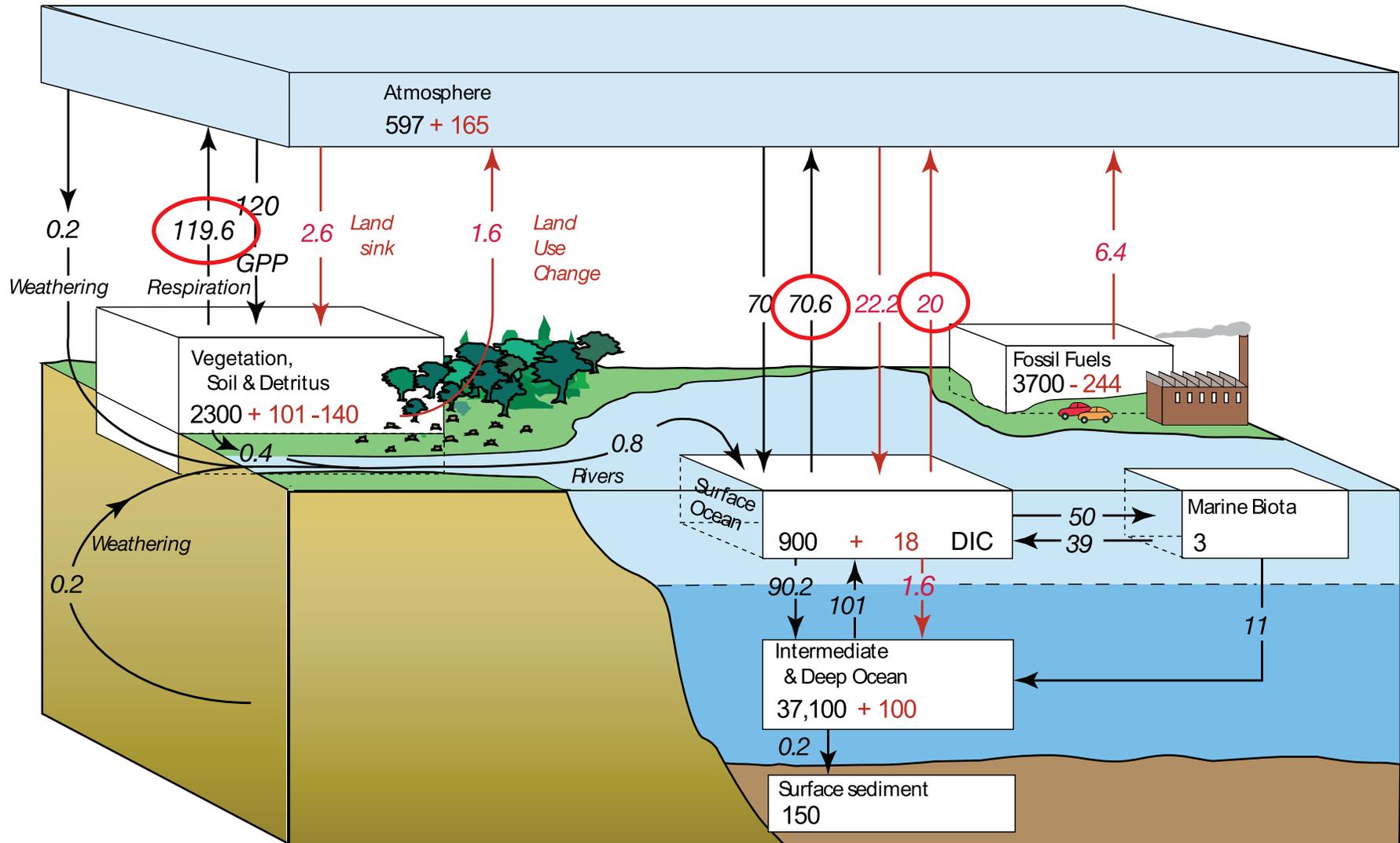


IPCC AR4, Chapter 7 (2007); after Sarmiento and Gruber (2002)

Large fluxes out of atmosphere suggest much shorter lifetime ~ 3.5 yr.

# THE GLOBAL CARBON CYCLE

Preindustrial and **anthropogenic perturbation (1990's)**  
 Stocks in upright type, Pg C; *flows in italic type*, Pg C yr<sup>-1</sup>

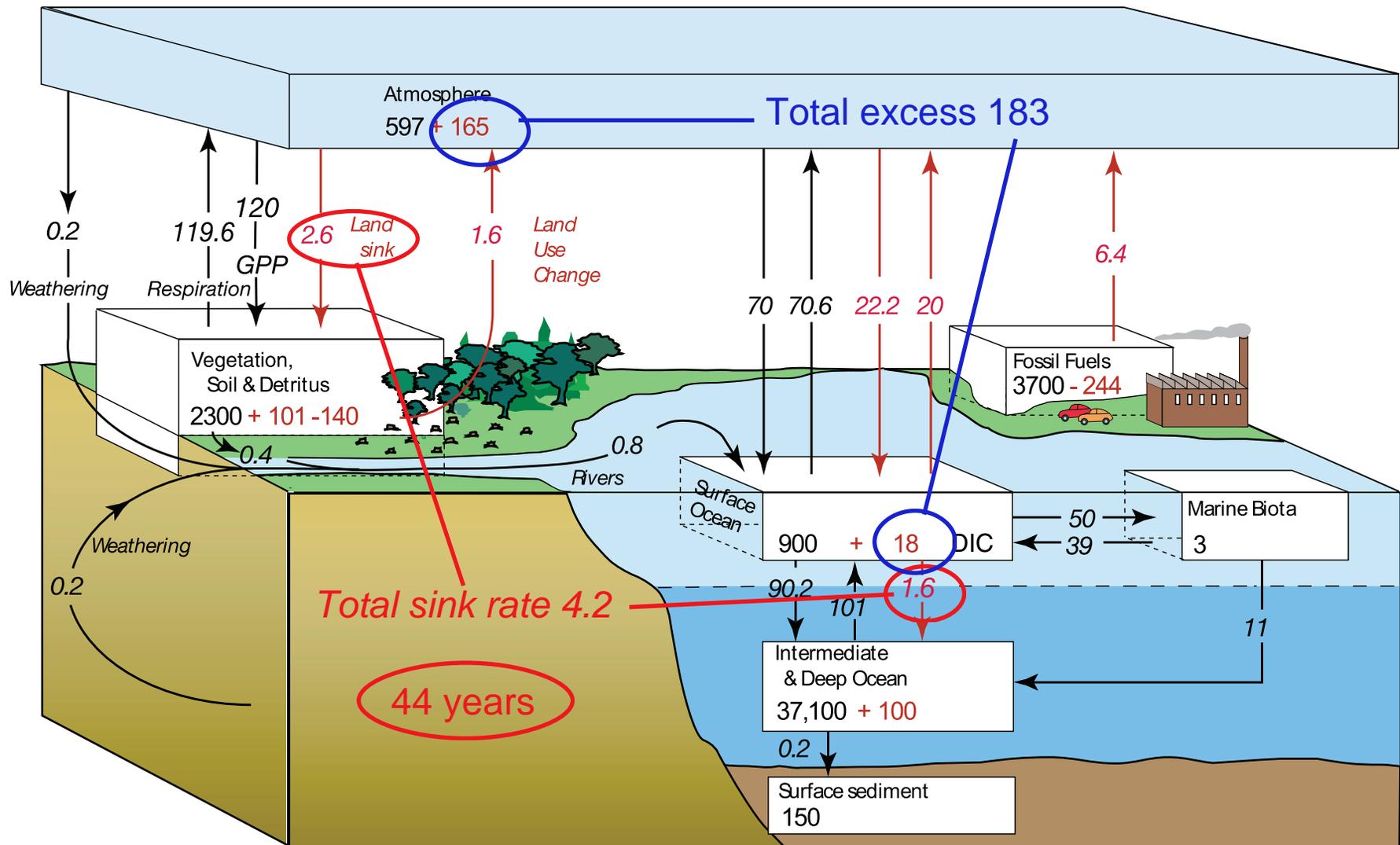


IPCC AR4, Chapter 7 (2007); after Sarmiento and Gruber (2002)

Paradox resolved: Large return flux indicates coupled reservoirs.

# THE GLOBAL CARBON CYCLE

Preindustrial and **anthropogenic perturbation (1990's)**  
 Stocks in upright type, Pg C; *flows in italic type*, Pg C yr<sup>-1</sup>

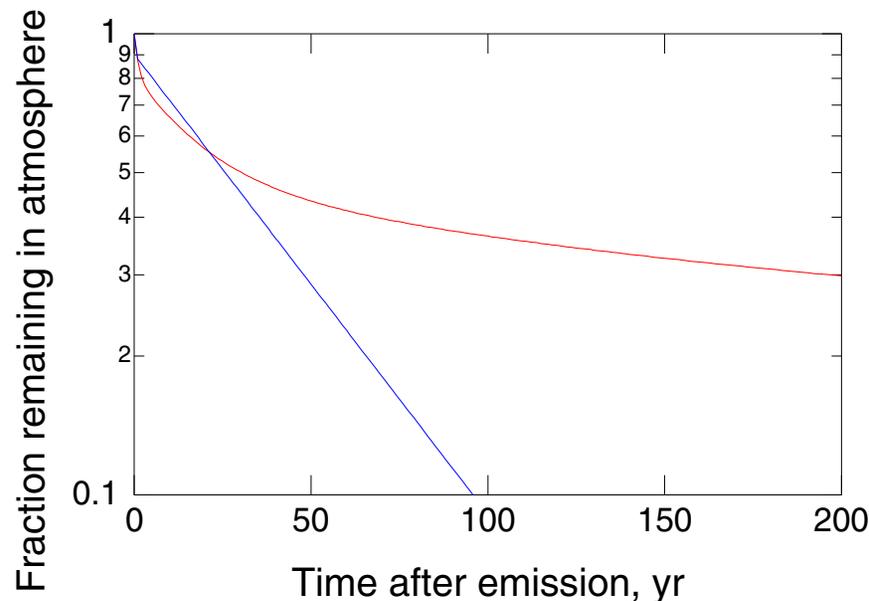
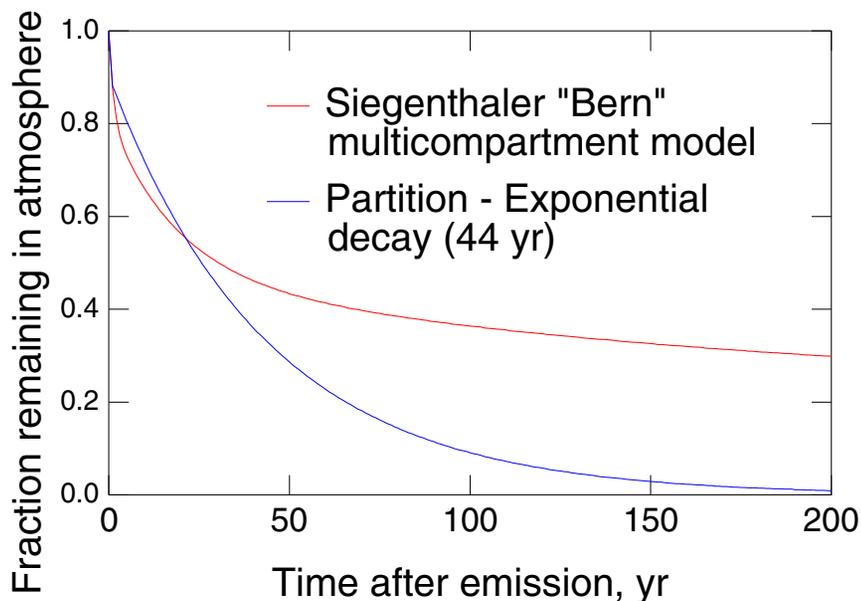


IPCC AR4, Chapter 7 (2007); after Sarmiento and Gruber (2002)

Turnover time calculated as (excess C)/(sink rate) is 44 yr.

# ALTERNATE VIEWS OF THE CO<sub>2</sub> IMPULSE RESPONSE FUNCTION

Comparison on linear and semilogarithmic plots

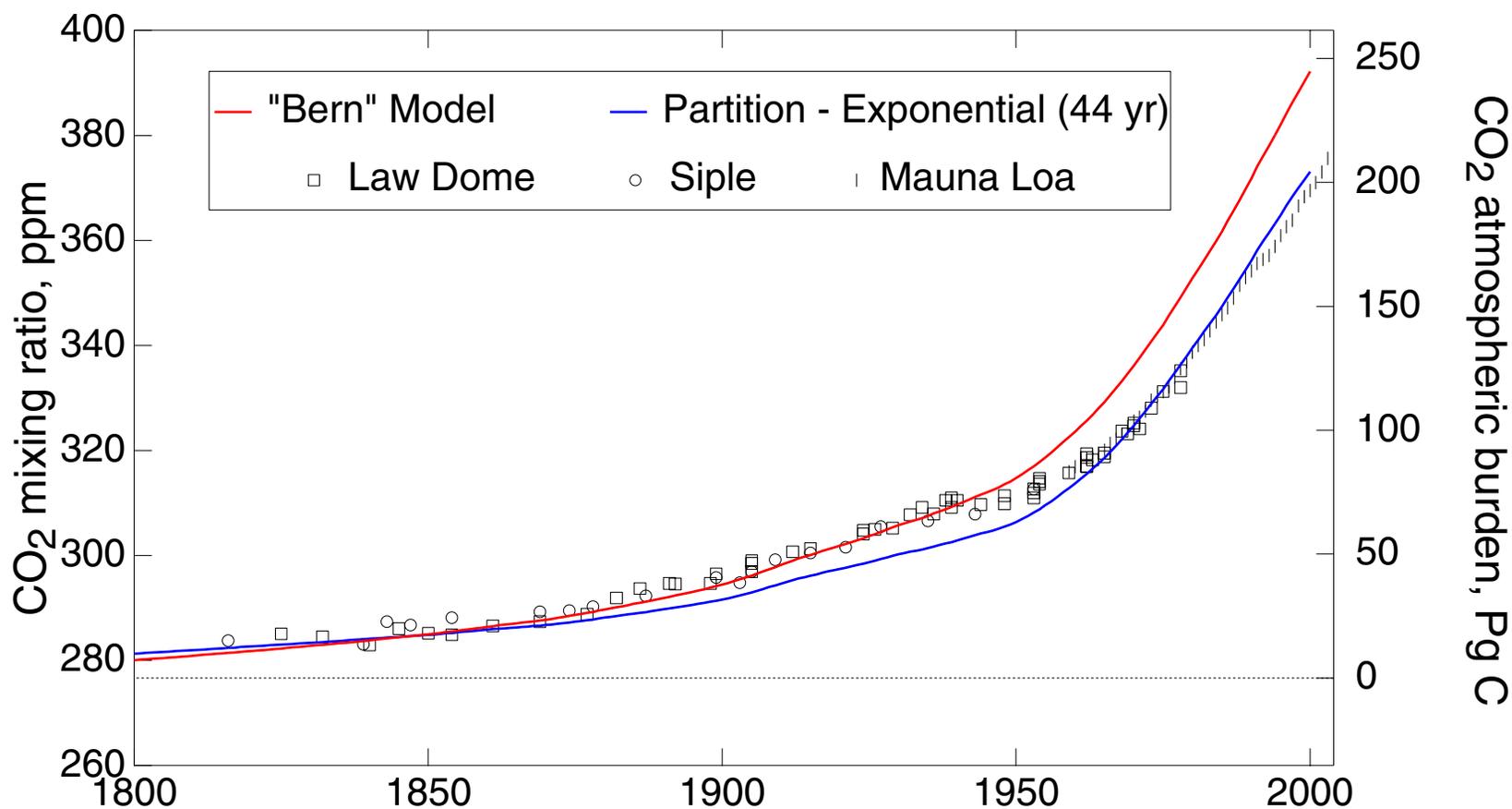


Bern model is basis for assertions that excess CO<sub>2</sub> is long lived in the atmosphere.

Response functions closely agree over initial 30 years but diverge greatly at long times.

# COMPARISON OF MODELED AND OBSERVED CO<sub>2</sub> MIXING RATIO

Fossil emissions from Marland; land-use emissions from Houghton



Both models reproduce observations given uncertainties in emissions.  
Long term impulse response is irrelevant for rapidly increasing emissions.  
Knowledge of the true impulse response is essential to energy planning.

# CHARACTERISTIC TIME OF EARTH'S CLIMATE SYSTEM

I add a watt per square meter to Earth's energy budget, and then

Earth warms up: Can you tell me when?

Incremental heating “in the pipeline” from excess CO<sub>2</sub> already in the atmosphere

Earth's climate sensitivity

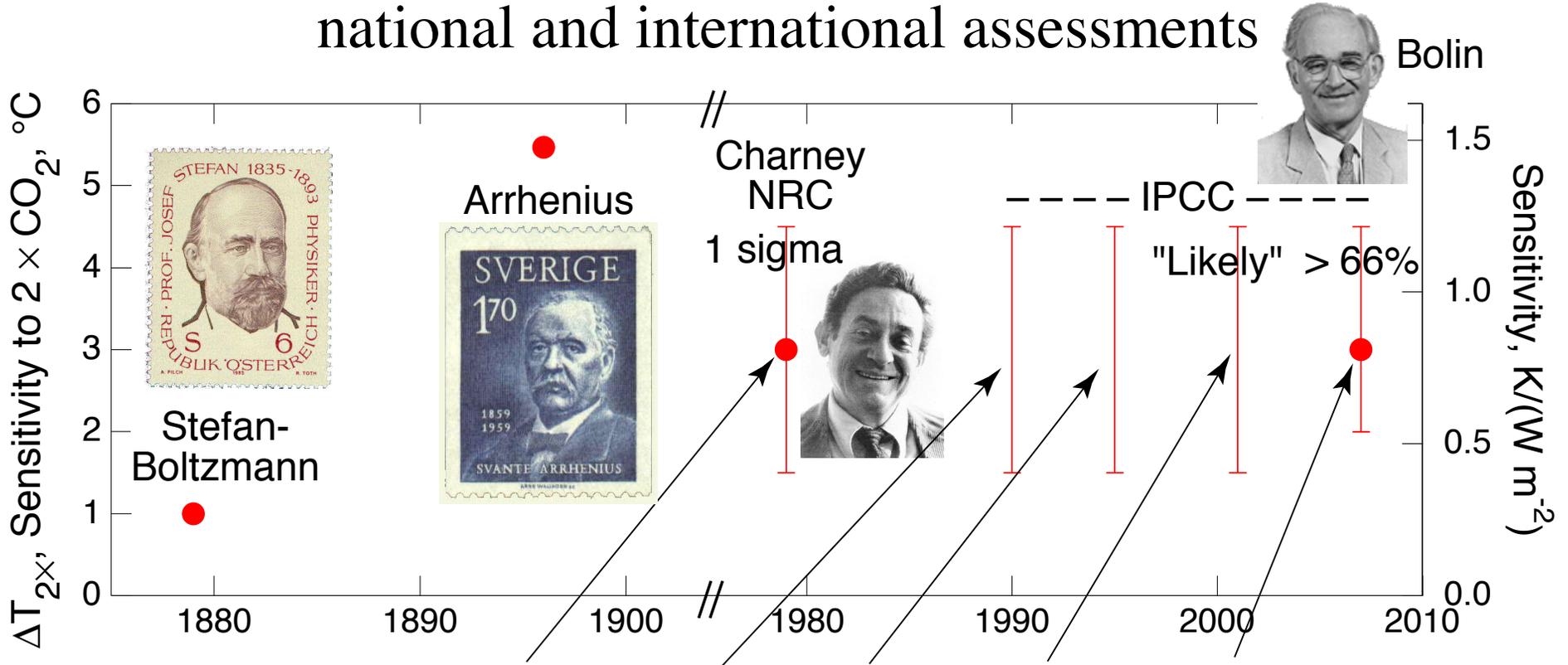
# APPROACH TO DETERMINE EARTH'S CLIMATE SENSITIVITY

Empirically determine heat capacity  $C$  and time constant of Earth's climate system from measurements over the instrumental period.

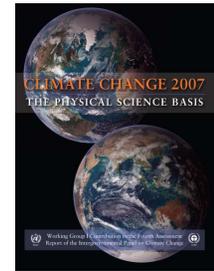
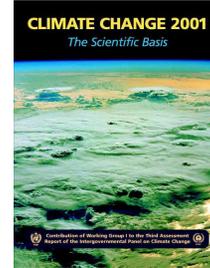
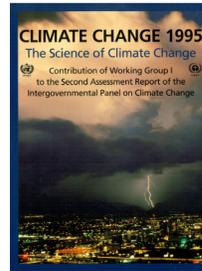
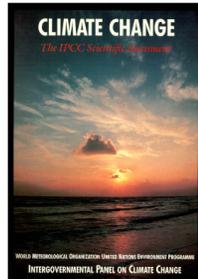
Evaluate sensitivity as  $\Delta T = \Delta Q / C$

# CLIMATE SENSITIVITY ESTIMATES THROUGH THE AGES

Estimates of central value and uncertainty range from major national and international assessments



**Carbon Dioxide and Climate:  
A Scientific Assessment**  
NATIONAL ACADEMY OF SCIENCES  
Washington, D.C. 1979



Despite extensive research, climate sensitivity remains *highly uncertain*.

# ENERGY BALANCE MODEL OF EARTH'S CLIMATE SYSTEM



Global energy balance:  $C \frac{dT_s}{dt} = \frac{dH}{dt} = Q - E = \gamma J - \varepsilon \sigma T_s^4$

$C$  is heat capacity coupled to climate system on relevant time scale

$T_s$  is global mean surface temperature       $H$  is global heat content

$Q$  is absorbed solar energy

$E$  is emitted longwave flux

$J$  is  $\frac{1}{4}$  solar constant

$\gamma$  is planetary co-albedo

$\sigma$  is Stefan-Boltzmann constant

$\varepsilon$  is effective emissivity

# ENERGY BALANCE MODEL OF EARTH'S CLIMATE SYSTEM

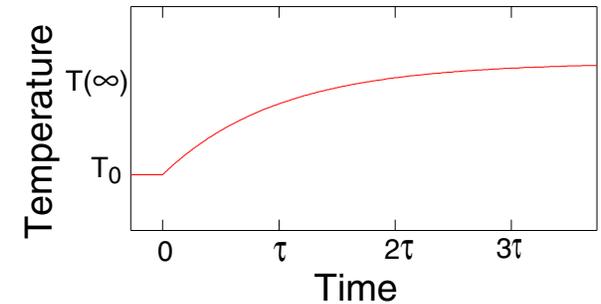


Apply step-function forcing:

$$F = \Delta(Q - E)$$

At “equilibrium”

$$\Delta T_s(\infty) = \lambda^{-1} F$$



$\lambda^{-1}$  is equilibrium climate sensitivity      Units:  $\text{K} / (\text{W m}^{-2})$

Time-dependence:

$$\Delta T_s(t) = \lambda^{-1} F (1 - e^{-t/\tau})$$

$\tau$  is climate system time constant       $\lambda^{-1} = \tau / C$

***One equation in three unknowns!***

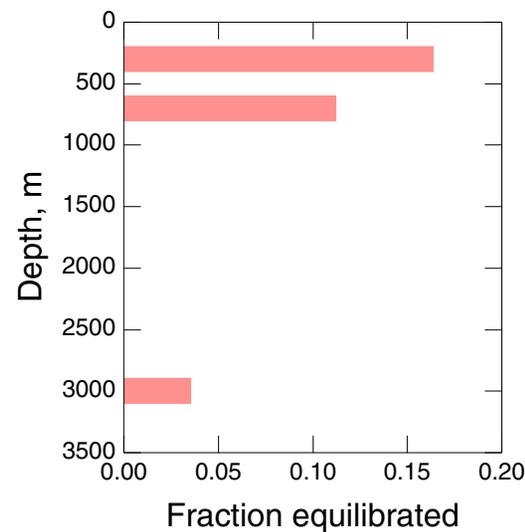
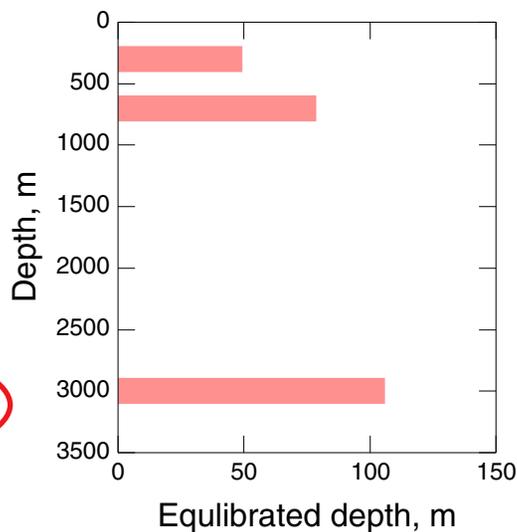
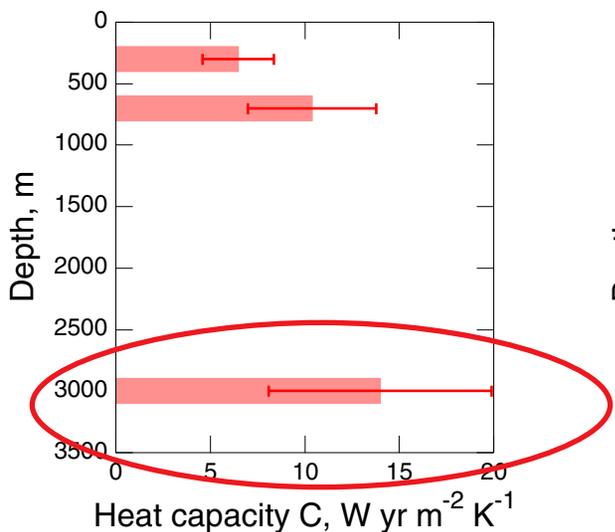
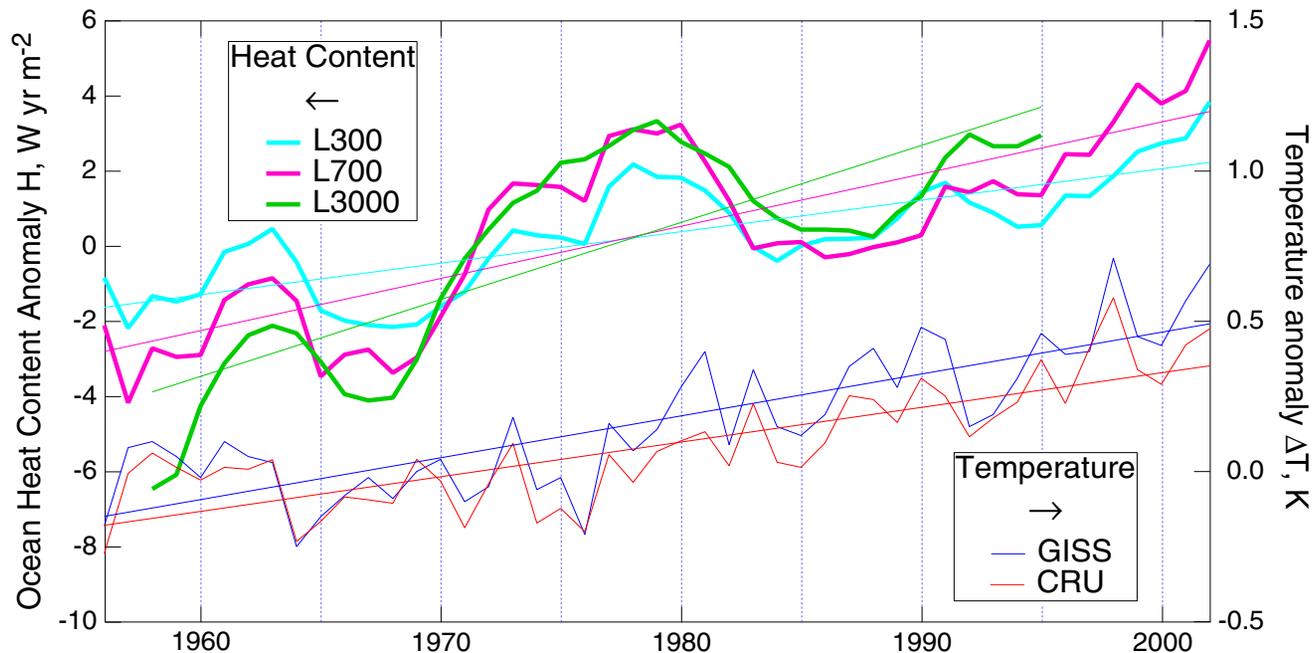
Approach: Determine  $C$  and  $\tau$  empirically; calculate sensitivity  $\lambda^{-1}$ .

# EMPIRICAL DETERMINATION OF OCEAN HEAT CAPACITY

$$C = \frac{dH / dt}{dT_s / dt}$$

Ocean heat content  
*H*: Levitus *et al.*,  
 GRL, 2005

Surface temperature  
*T<sub>s</sub>*: GISS, CRU

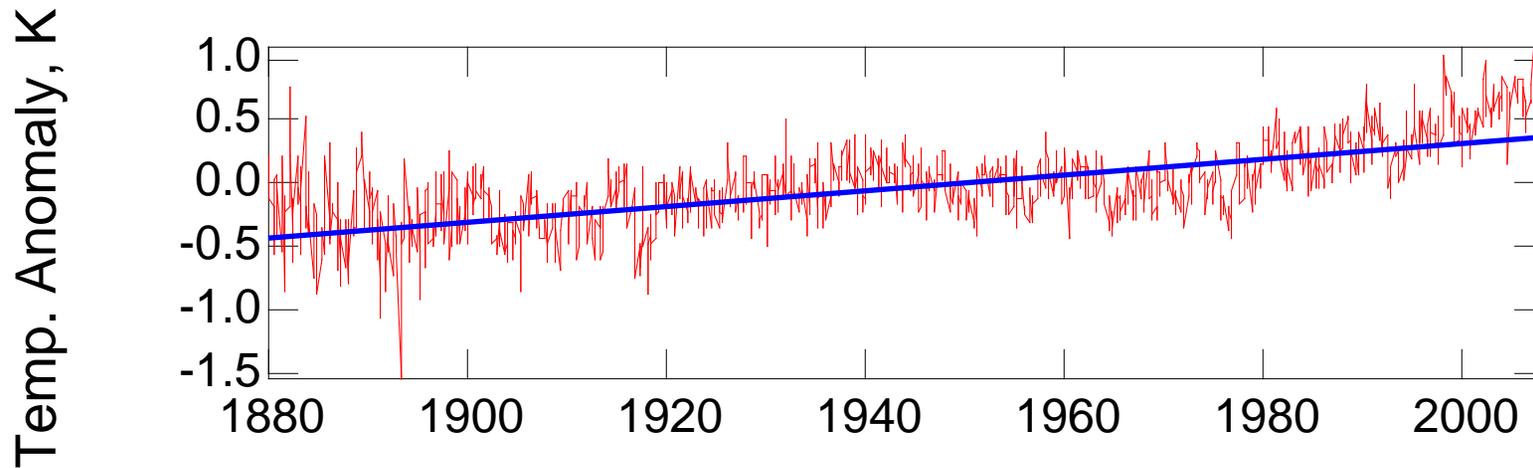


- ~50% of heat capacity is between surface and 300 m.
- Other heat sinks raise global heat capacity to  $17 \pm 7 \text{ W yr m}^{-2} \text{ K}^{-1}$ .

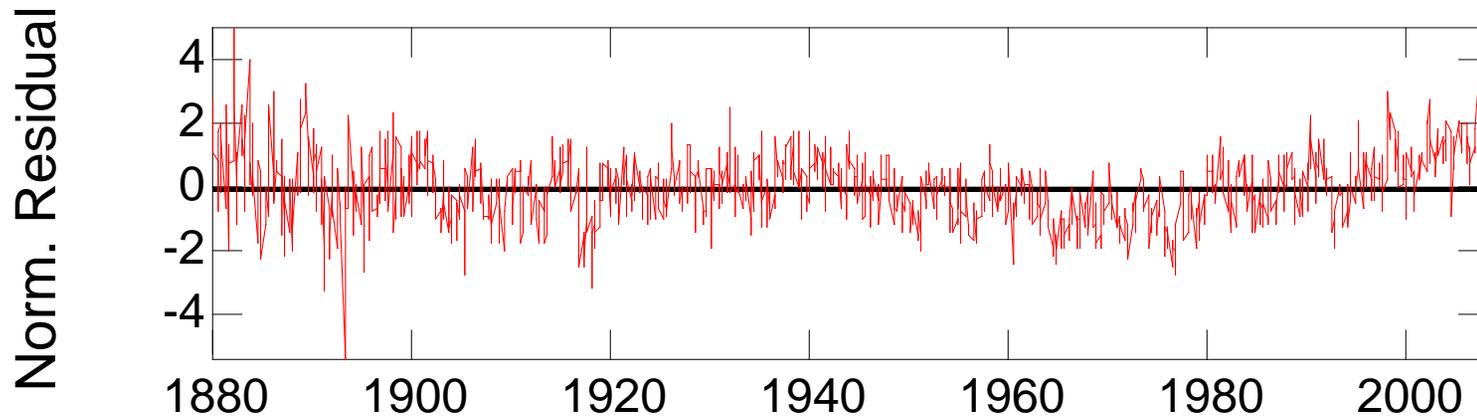
# TIME CONSTANT OF EARTH'S CLIMATE SYSTEM

## Determination from autocorrelation of time series

**Input:** Monthly global-mean surface temperature anomaly  $T_s$



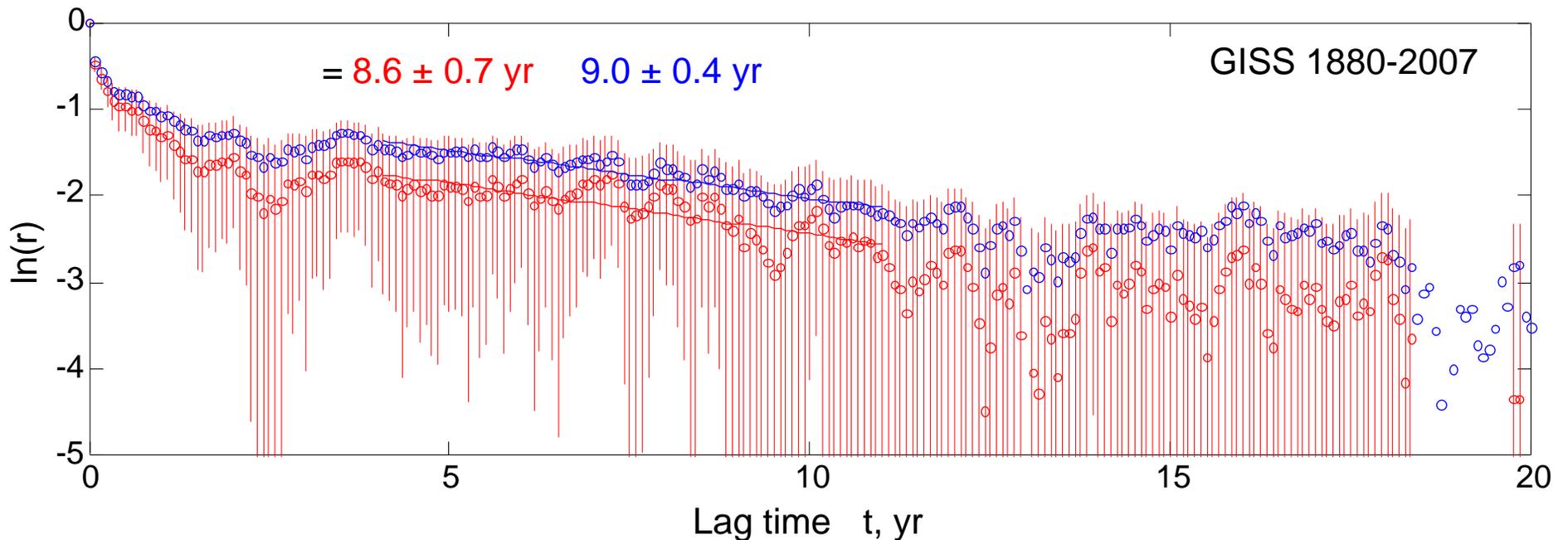
Remove long term trend; plot the residuals:



# TIME CONSTANT OF EARTH'S CLIMATE SYSTEM

Determination from autocorrelation of time series (*cont'd*)

Evaluate *climate system time constant* as  $\tau = (d \ln r(t) / d t)^{-1}$   
*Correct for short duration of time series.*



Summary (multiple data sets):

Climate system time constant is  $8.5 \pm 2.5$  years

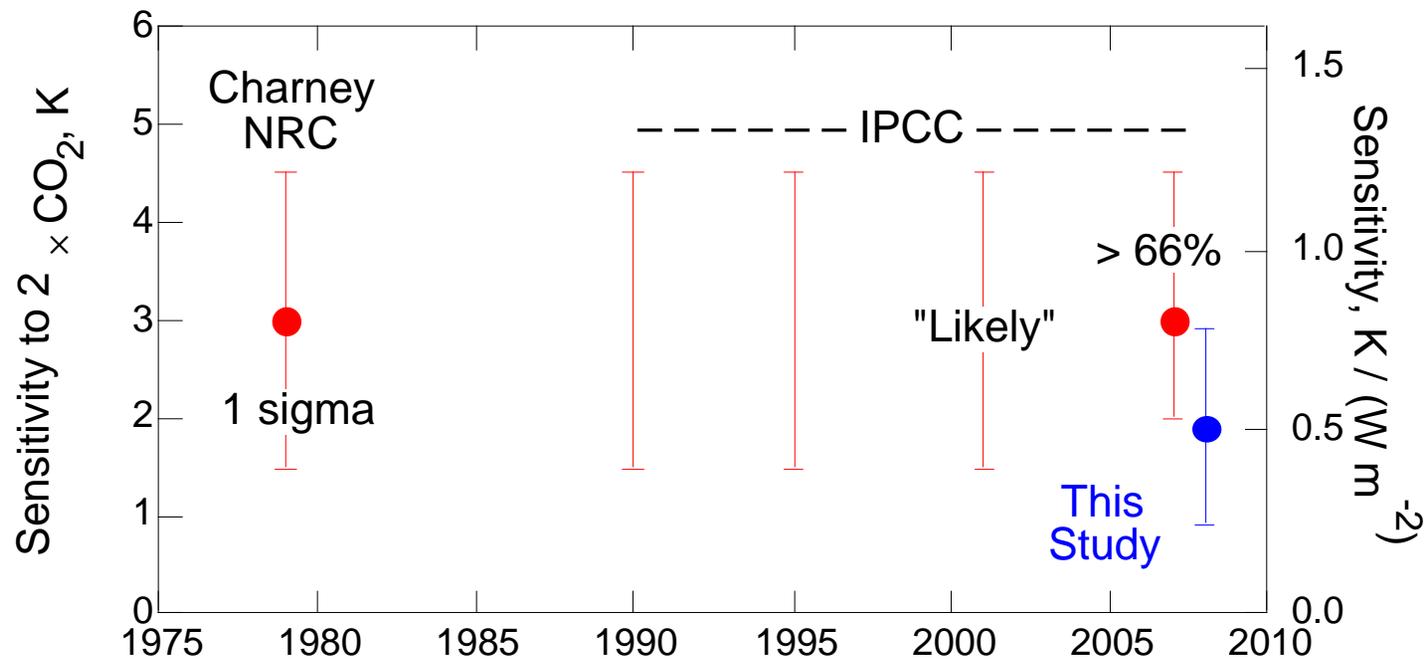
# IMPLICATIONS

*Climate system time constant* is  $8.5 \pm 2.5$  years.

Short time constant implies *little further heating “in the pipeline”* from present greenhouse gases.

Earth's *climate sensitivity* is  $0.51 \pm 0.26$  K / (W m<sup>-2</sup>).

*CO<sub>2</sub> doubling temperature*  $T_{2\times} = 1.9 \pm 1.0$  K.



IT'S ABOUT TIME



THANK YOU, HENNING