

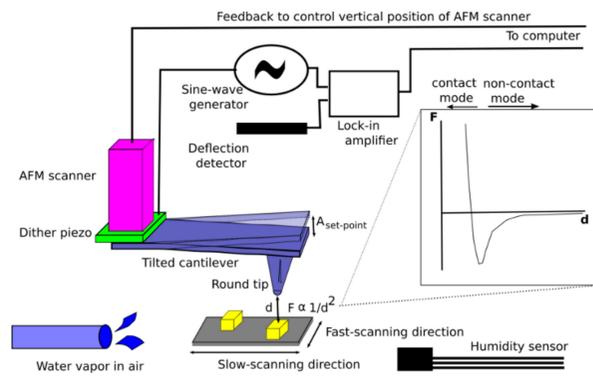
## Abstract

- The response of NaCl nanoparticles to relative humidity (RH) was characterized by measuring the height of particles deposited on a prepared hydrophobic surface with **non-contact environmental atomic force microscopy** (e-AFM).
- Cubic NaCl nanoparticles (35–150 nm wide) **reversibly take up water** with increasing RH to form a liquid-like surface layer 2–4 nm thick at RH **well below the bulk deliquescence point** of 75% at 20 °C. Measurable uptake begins at 70% RH.
- The maximum **thickness of the layer increases with increasing RH** for a given particle size, and for a given RH, **increases with increasing particle size**.
- The liquid-like behavior of the layer is indicated by a **reversible “rounding” at the tops of the particles**, where the ratio of particle height to radius of curvature increases from zero (flat top) at 68% RH to 0.7 at 74% RH.
- These observations suggest that a reorganization of mass occurs on the solid NaCl nanocrystal, and that the deliquescence of NaCl nanoparticles between 70% and 75% RH is **more complex than an abrupt first-order phase transition**.
- Theoretical treatments of the phase transition should account for the presence of a liquid-like layer prior to deliquescence, and for its RH-dependent thickness.

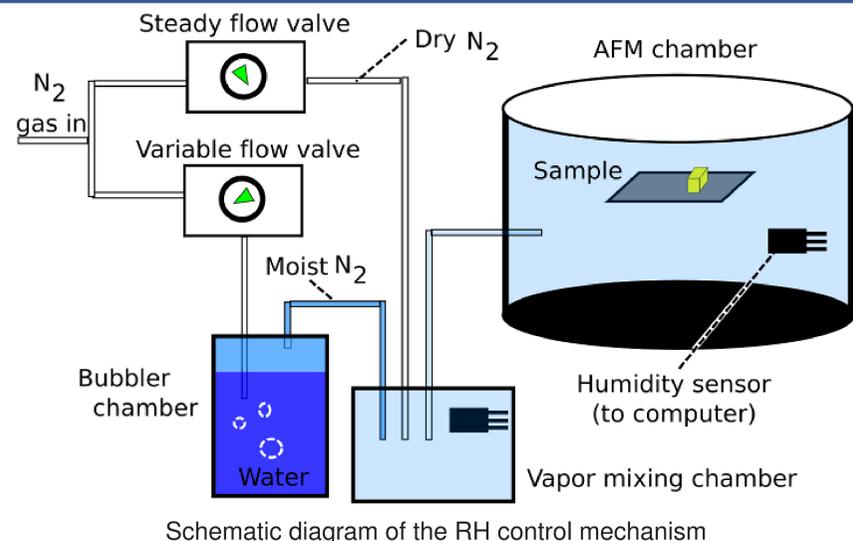
## Non-contact environmental AFM

### Nanoimaging at variable relative humidity

- 5–10 nm gap maintained
- No transfer of material between tip and sample
- Sensitivity to changes in height of ~1 nanometer
- 2% accuracy of RH measurement

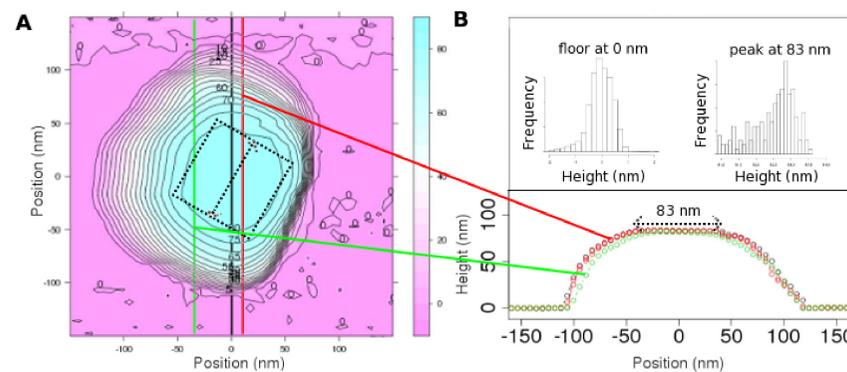


## Controlling relative humidity (RH)



## Design and Conduct of Experiments

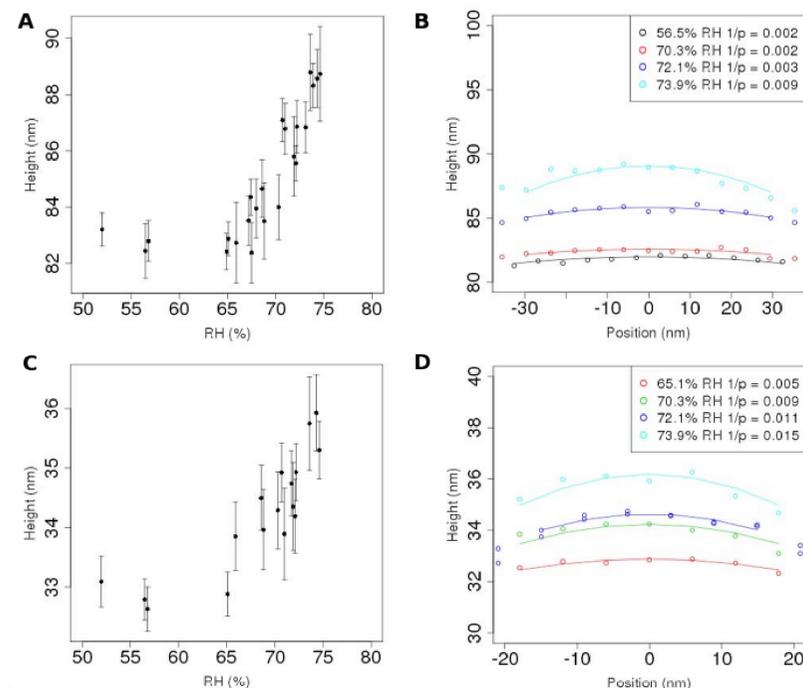
### AFM imaging and data analysis



- Prepare moderately hydrophobic surfaces ( $\angle_{\text{water, static contact}} \sim 75^\circ$ )
- Synthesize cubic particles and deposit dry onto substrate
- Measure apparent height of particle relative to substrate vs. RH and position
- Correct for tip effects, especially convolution; focus on center of flat top

## Results

### Prewetting of NaCl at RH below deliquescence point



Error-bars denote standard deviation of peak height. Points shown without error-bars are single measurements.

- Reversible** formation of liquid-like layer 2–5 nm thick
- Statistically significant onset of **adlayer** at **~70% RH**
- Distinct rounding** of initially flat tops with increasing RH

## Theory

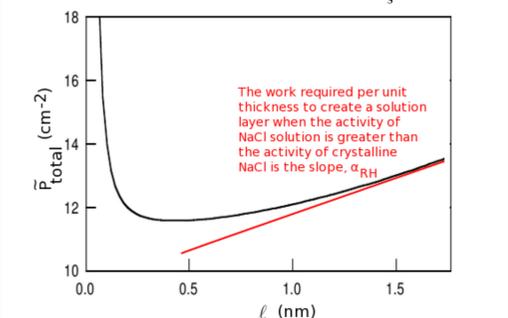
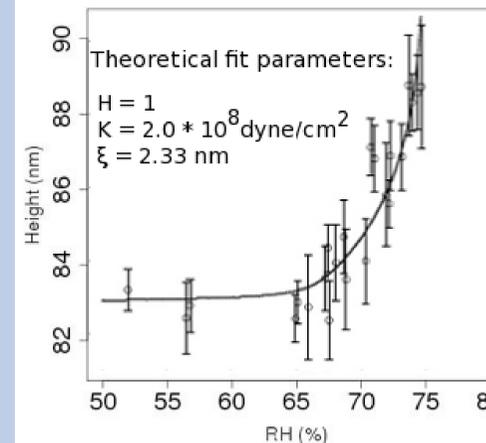
### A model for wetting includes:

- Free-energy to form layer  $\ell$  thick (1)
- van der Waals interaction (2)
- Approximation of disjoining pressure (3)

$$\tilde{P}_{\text{capillary approx}}(\ell) = \alpha_{RH} \ell \quad (1)$$

$$\tilde{P}_{\text{van der Waals}}(\ell) = \frac{H}{12\pi\ell^2} \quad (2)$$

$$\tilde{P}_{\text{disjoining}}(\ell) = \frac{K\xi}{KT} \exp\left(-\frac{\ell}{\xi}\right) \quad (3)$$



The height  $\ell$  of the equilibrium layer minimizes the total energy,  $KT\tilde{P}_{\text{total}}$  per unit area, at each RH. The value used for pressure  $K$  is twice that of Kuni *et al.* (1996).

## Summary and Conclusions

- e-AFM can image single nanoparticles** over a range of RH, and can detect the onset of water-uptake at lower RH than is observed by polarization force microscopy or by environmental TEM.
- Reversible formation of a liquid-like layer** occurs at **RH below bulk deliquescence**.
- The liquid-like layer can be modeled by a continuum model **using parameters within a range used by other investigators**. The thickness of the modeled layer is **insensitive to the estimate of the layer's Hamaker constant, H**.
- The presence of moisture may have **implications for light scattering**, and for the **restructuring of atmospheric aerosol particles** with morphologies that are important to Earth's radiation balance and to climate change.

## References

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- F. M. Kuni, A. K. Shchekin, A. I. Rusanov, and B. Widom. Role of surface forces in heterogeneous nucleation on wettable nuclei. *Adv. Colloid Interface Sci.*, 65:71–124, 1996.
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